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State #20

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Rates of

Reactions - Part

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States of Matter

(gas, liquid,

and solid) and

Physical vs.

Chemical

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1 - Part 3 **The**

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Constant

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STUDY. PLAY.

energy. the

capacity to do

work or produce

heat. law of

conservation of

energy. states

that in any

chemical

reaction or

physical

process, energy

can be converted

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from one form to another, but it neither created nor destroyed.

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...

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change chapter

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Terms in this

set (75) boiling

point elevation.

The temperature

difference

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boiling point
and a pure
solvent's
boiling point.
Brownian motion.

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*A 155-g sample
of an unknown
substance was
heated from ...*

Figure 15.4.1
Energy Changes
Accompanying the
Thermite
Reaction Because
enthalpy is a
state function,
the overall
enthalpy change

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of 2 mol of
Al(s) with 1 mol
of Fe₂O₃(s)

is -851.1 kJ,

whether the

reaction occurs

in a single step

(ΔH 4, shown on

the left) or in

three

hypothetical

steps (shown on

the right) that

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experts! The
mass of the lead
(density 11.4
 g/cm^3) having
a volume equal
to the 15.0 g
of mercury
(density 13.6
 g/cm^3) needs
to be

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