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Unit 6 - Test

(Chemical

Compound) In

chemistry, a

compound refers to

an entity that

consists of two or

more atoms, with

at least two of

those atoms from

different chemical

elements. These

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atoms associate
via chemical
bonds.

Unit 6 -
Test(Chemical
Compound) -
ProProfs Quiz
The lines/curves
indicate the
maximun amount
of solute which can
be dissolved in
100g of water at

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each temperature.
We call this a saturated solution.. Any amount above the curve is said to be supersaturated; there is more solute than the solvent can hold, therefore the excess precipitates out/settles to the bottom.. Any amount below the

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line is called
unsaturated,
meaning more
solute can be ...

Unit 6 | chemistry

UNIT 6:

RADIOACTIVITY

AND NUCLEAR

CHEMISTRY 5. A

skeleton of a

mammoth was

analysed and found

to contain 21.2% of

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the carbon-14
found in living
bone. The half-life
of carbon-14 is
5730 years. What
is the most likely
year of the
mammoth's death?
Use $\log(N_i/N) =$
 $t \log_2 t_{1/2}$ $N_i/N =$
 $100/21.2$, $t_{1/2} =$
5730 years; hence
find t This year - t
= year of death

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Flashcards | Quizlet

UNIT 6 – REDOX

REACTIONS 6 □ The

oxidation number

of an atom is the

charge that would

exist on an

individual atom if

the bonding were

completely ionic □

In simple ions, the

oxidation number

of the atom is the

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Charge on the ion: -
Na⁺, K⁺, H⁺ all
have an oxidation
number of +1 -
Mg²⁺, Ca²⁺, Pb²⁺
all have an
oxidation number
of +2 - Cl⁻, Br⁻, I⁻ all
have an oxidation
number of -1

UNIT 6 REDOX
REACTIONS - A-
Level Chemistry

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study of the

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composition,
behaviour and
properties of
matter, and of the
elements of the
Earth and its
atmosphere.

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