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$pK_{BH} + (1) = 2,38$ (25°C,) $pK_a (1) = 9,39$ (25°C,) . (II) . 1,4- -2- .

I-

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Empirical formula (Hill's system for organic substances): C 16 H 36 IN Structural formula as text: (C4H9)4NI Molar/atomic mass: 369.368
Melting point (°C): 147 Solubility (g/100 g of solvent): 1-butanol: 62.28 (25°C) water: soluble methanol: 256.5 (25°C) chloroform: sparingly soluble ethanol: soluble . References:

tetrabutylammonium iodide

Empirical formula (Hill's system for organic substances): C 18 H 35 AgO 2 Structural formula as text: CH3(CH2)16COOAg ... Dictionary of organic compounds. - Vol. 3, naphthacarbazole - zygadenine. - ?.: ???, 1949. - pp. 637;

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Empirical formula (Hill's system for organic substances): C 18 H 16 O 7 Molar/atomic mass: 344.33 Melting point (°C): 196 Solubility (g/100 g of solvent): water: insoluble diethyl ether: soluble chloroform: very soluble ethanol: soluble . Dissociation: pK a (1) = 4.4 (25°C, water) pK a (2) = 8.8 (25°C, water)

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