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problems. Show your work and include units for full credit. 1. Calcium hydroxide (["slaked lime"]) and sulfuric acid react to produce calcium sulfate and water according to the following balanced equation:  $\text{Ca(OH)}_2 \text{ (aq)} + \text{H}_2\text{SO}_4 \text{ (aq)} \rightarrow \text{CaSO}_4 \text{ (s)} + 2\text{H}_2\text{O (l)}$

Molarity and Stoichiometry

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Worksheet With Answers

Answer the following stoichiometry-related questions: 12) Write the balanced equation for the reaction of acetic acid with aluminum hydroxide to form water and aluminum acetate: 13) Using the equation from problem #12, determine the mass of aluminum acetate that can be made if I do this reaction with 125 grams

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of acetic acid. 16 Best Images Of Wave  
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