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Periodic Trends 1 Model 1 – Main Group Elements 1 H 37 1312 2.1 3
Li 4 Be 152 520 1.0 11 Na 186 496 0.9 19 K 160 738 1.2 20 Ca 227 404
0.8 143 578 1.5 31 Ga 8 O 9 F 77 1086 2.5 14 Si 71 1402 3.0 15 P 66
1314 3.5 16 S 71 1681 4.0 17 Cl 70 2081 N/A 18 Ar 117 786 1.8 32 Ge
122 558 1.7 7 N 83 801 2.0 13 Al 197 550 1.0 6 C 112 900 1.5 12 Mg 5 B
2 He 31 2372 N/A 10 Ne 115 1011 2.1 33 As 123 709 1 ...

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Rank the following elements from smallest to largest electronegativity based on the trends you have discovered thus far in the periodic table: barium (atomic number 56), bromine (atomic number 35), and iron (atomic number 26).

15 Periodic Trends-S

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electron from an atom a Using your knowledge of Coulombic attraction, explain why ionization—removing an electron from an atom—takes energy. Which takes more energy, removing ...

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Periodic trends are specific patterns that are present in the periodic table that illustrate different aspects of a certain element, including its size and its electronic properties. Major periodic trends include: electronegativity , ionization energy , electron affinity , atomic radius , melting point, and metallic character .

Periodic Trends - Chemistry LibreTexts

Periodic Trends . 6. Locate the numbers in Model 1 that represent the ionization energy. The ionization energy is the amount of energy needed to remove an electron from an atom. a. Using your knowledge of Coulombic attraction, explain why ionization—removing an

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October 17 Periodic Table Scavenger Hunt and Key Page 1 Page 2
Thursday: October 18 Periodic Trend Quiz Friday: October 19
Periodic Table Short Answer Review Key Page 1 Page 2 Monday:
October 22 Test: Short Answer

Periodicity (Periodic Trends) - Weebly

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Students will be using experimental data to infer the periodic trends for atomic radius, first ionization energy and electronegativity. Patterns
Cross Cutting Concept : Graphs, charts, and images can be used to identify patterns in data. Science and Engineering Practice 2: Developing and using models .

Ninth grade Lesson Periodic Trends- Reciprocal Style Day 1
trends in the periodic table. § 1. Find the elements listed in the periodic table. § 2. Put the elements in order and determine the trend.
§ Sodium, Na. § Because atomic radii decreases across a period. § Of the elements calcium, Ca, beryllium, Be, barium, Ba, and strontium, Sr, which has the largest atomic radius? Explain your answer in ...

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