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Solids, Liquids, and

Gases Kinetic Theory

SECTION States of

Matter If you don ' t

finish lunch quickly,

you ' ll be late for

practice. The soup is

boiling on the stove. You

hastily pour the soup into

the bowl, but now it ' s

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too hot to eat. You add an ice cube and stir. The soup's temperature drops—now you can eat it without burn-

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Chapter 16: Solids,
Liquids, and Gases
Physical Science Chapter
16 The Periodic Table
Parts of an Atom • An
atom consists of a
nucleus surrounded by
one or more electrons •

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Atoms are electrically neutral w/ the same number of protons as electrons. • Majority of the atom is empty space. If nucleus were the size of a pencil eraser, the closest electron would be 100 yards away!

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how heat and work

transfer energy. 16.1.2

Relate thermal energy to

the motion of particles

that make up a material.

16.1.3 Relate temperature

to thermal energy and to

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thermal expansion.

16.1.4 Calculate thermal energy, temperature change, or mass using the specific heat equation.

16.1.5 Describe how a calorimeter

Section 16.1 16.1

Thermal Energy and
Matter

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