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e4 Mathematics for Physical Chemistry

thesineandcosinetotheappropriatenumberofdigits. (31) $2\pi \text{ rad } 360 = 0.54 \text{ rad}$ $\sin(30.5^\circ) = 0.5075$ $\sin(31.5^\circ) = 0.5225$ $\sin(31^\circ) = 0.51$ $\cos(30.5^\circ) = 0.86163$ $\cos(31.5^\circ) = 0.85264$ $\cos(31^\circ) = 0.86$ 15. Some elementary chemistry textbooks give the value of R, the ideal gas constant, as $0.0821 \text{ l atm K}^{-1} \text{ mol}^{-1}$. a.

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Solutions to exercises 1B.1(b) The mean speed is $8RT/v_{\text{mean}}$ M The mean translational kinetic energy is $m \cdot 3RT/2$ M The ratios of species 1 to species 2 at the same temperature are $E_k/v_{\text{mean},1}$ $v_{\text{mean},2}$ (i) $1/2$ $m v^2$ M_1 $v_{\text{mean},H}$ $1/2$ $m v^2$ and $2/v_{\text{mean},Hg}$ 200.6 4.003 $1/2$ 2 $m v_{\text{rms}}$ E_k 1 E_k 2 7.079 (ii) The mean translation kinetic energy is independent of molecular mass and depends upon temperature alone!

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